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*Olive leaves: From waste to active in pharmaceutical micro and nanoformulations*

European project aimed at valorizing waste from olive oil production, to make products of pharmaceutical, cosmetic and food interest



## OLIVE LEAF MULTI-PRODUCT CASCADE BASED BIOREFINERY

From an under-used biomass in the primary sector to tailor-made solutions for high added value international market applications

JULY 1<sup>ST</sup> 2021

JUNE 30<sup>TH</sup> 2024

36 months

Coordinated by: NATAc GROUP (ES)

Overall budget: € 5.687.060

Project partners: 16

Countries: 9 EU countries

Grant Agreement: 101023256



# 16 PARTNERS & 9 EUROPEAN COUNTRIES INVOLVED



# OLEAF4VALUE covers the whole value chain

1. Olive biomass: Smart dynAmic Multi-valorisation-route BIOfinery (SAMBIO) for the cascade valorisation of the olive leaf biomass according to its physicochemical composition, particularly modulated by specific pretreatments to produce target products

3. Applications: Enzymatic biotransformation and nanoencapsulation technologies will be applied to develop tailor made prototypes according to end user market needs from high value sectors: food, feed, health, cosmetic, pharma and chemical industries.

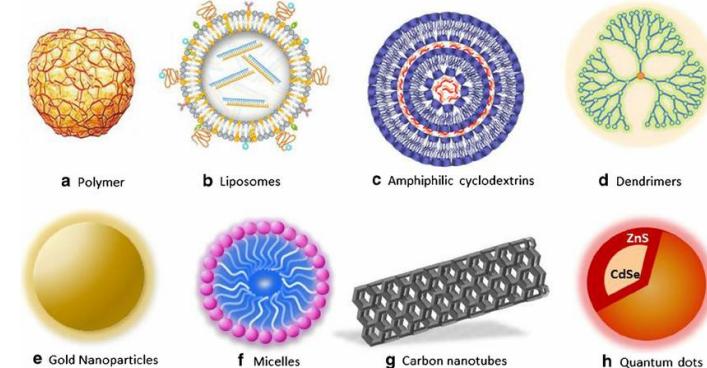
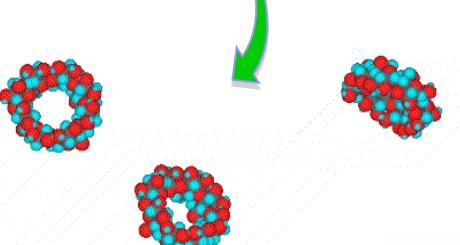
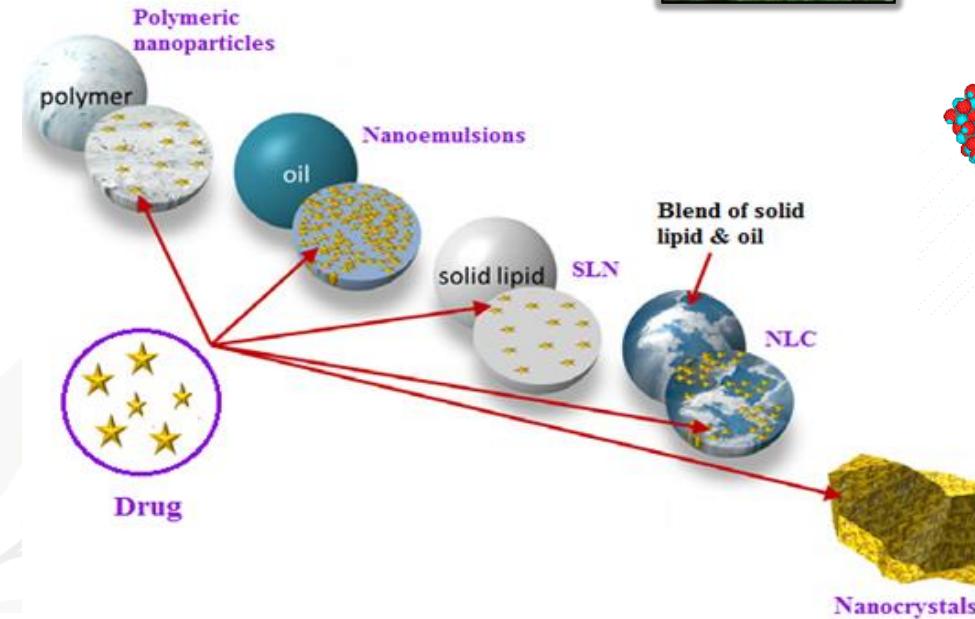
2. Biorefining: Advanced green extraction and isolation technologies are used to sequentially separate all fractions and compounds of value, with a zero-waste approach.

4. Market: Large companies from these sectors within the consortium will guarantee a good market-oriented approach throughout the project.

# University of Florence (UNIFI)



RI7464405



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- Increase solubility
- Protection against chemical and biological degradation processes
- Increased permeation and cellular internalization
- Modified release
- Site-specific delivery through superficial modifications
- Reduction of side effects deriving from systemic distribution

# Role of UNIFI in OLEA4VALUE project



- Micro/Nanoformulations to improve technological performance of olive bioactive compounds.
- *In vitro* evaluation of influence of optimized formulations on solubility and permeability of olive bioactive compounds and olive leaf fractions.

**Microemulsions**

**Polymeric micelles**

**Solid dispersions**

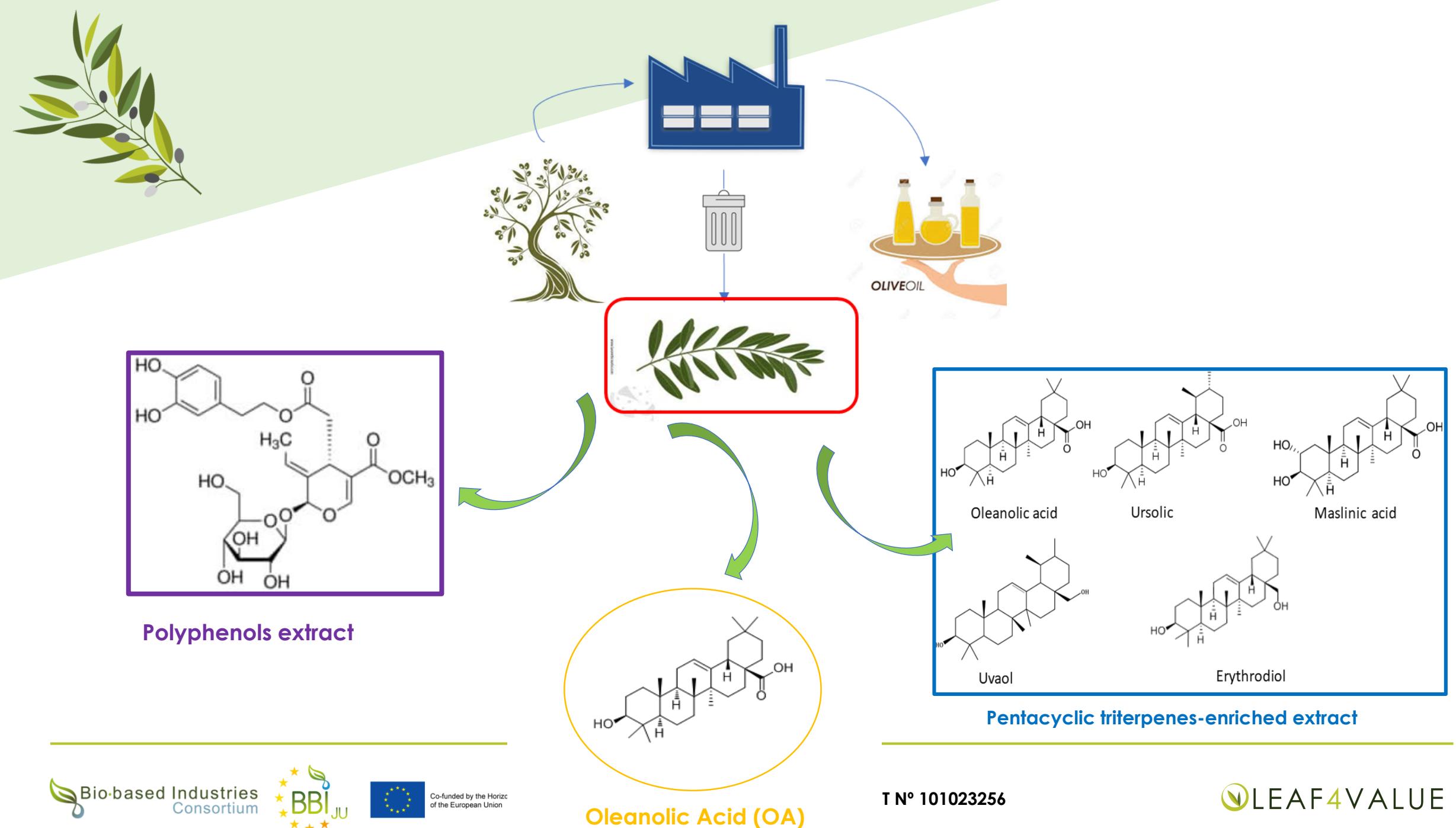
**Co-ground products**

***In vitro Solubility test***

**Dissolution test**

**PAMPA and Caco-2 cells**

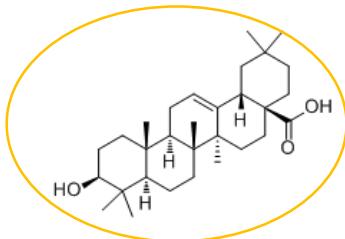




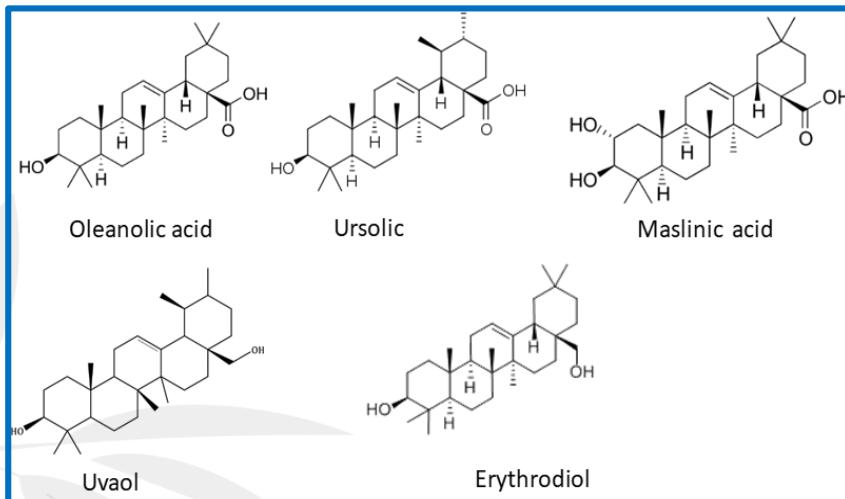
# Microemulsions

Systems of water, oil, surfactant which form an optically isotropic and thermodynamically stable solution

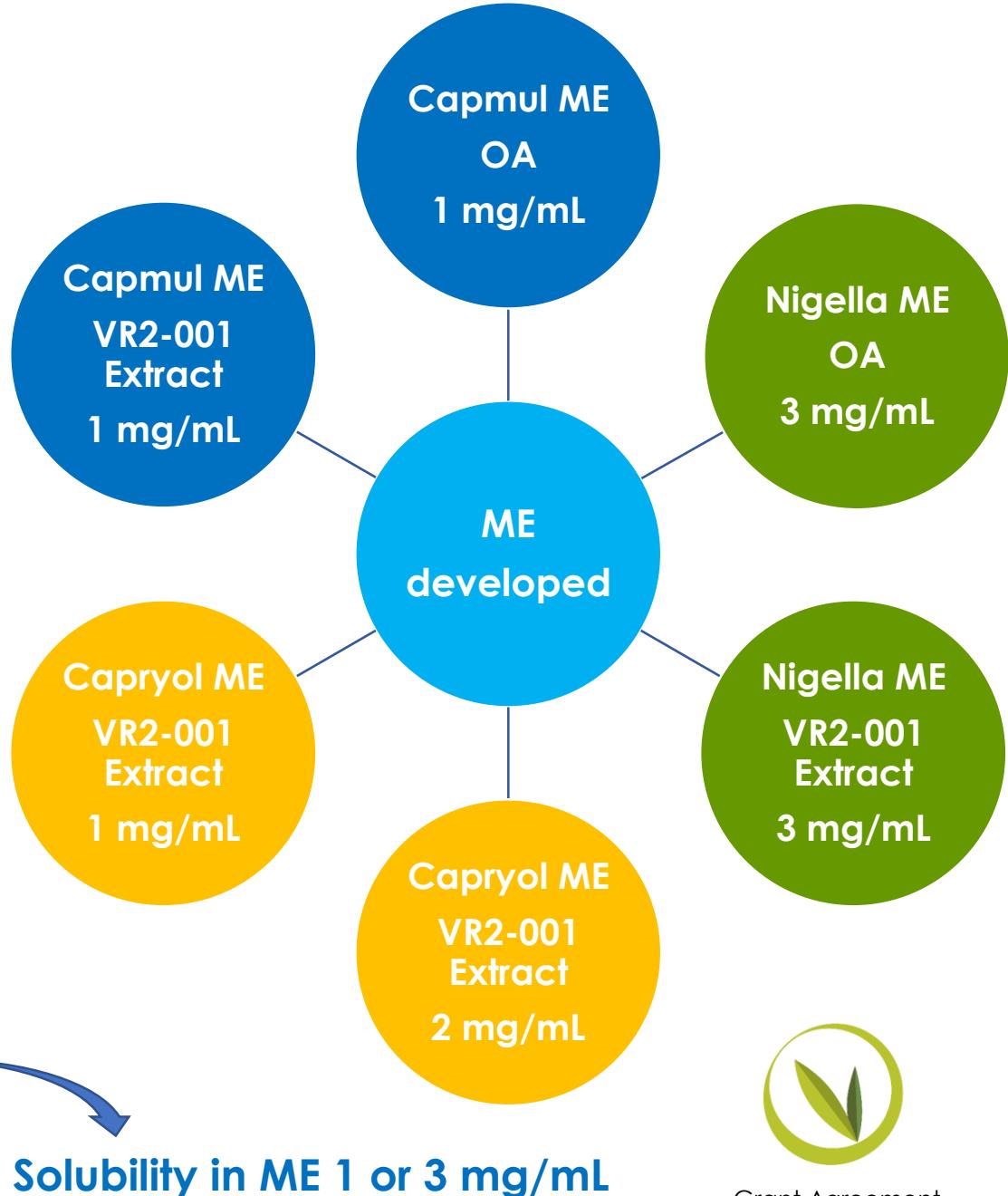
Oleanolic Acid (OA)



Pentacyclic triterpenes-enriched extract (TTP70)



Solubility of OA in water < 1 µg/mL  
Solubility of Extract in water: TTPs ~ 7 µg/mL



# MEs' composition

Capmul PG8 5.95 % w/w  
Transcutol HP 37.20 % w/w  
Tween 20 15.98 % w/w  
Water 40.48 % w/w  
  
1 mg/mL of OA  
1 mg/mL of NATAc extract

Transcutol HP: Diethylene glycol monoethyl ether  
Capmul PG8: Propylene glycol monocaprylate

Transcutol HP 36% w/w  
TPGS 4% w/w  
Capryol 90 5% w/w  
Water 55% w/w  
  
1 mg/mL of NATAc extract

TPGS: D-a-Tocopherol polyethylene glycol 1000 succinate  
Capryol® 90 - Propylene glycol monocaprylate

Nigella oil 2% w/w  
IPM 2% w/w  
Transcutol 30% w/w  
Cremophor EL 16% w/w  
Water 50% w/w  
  
3 mg/mL of OA  
3 mg/mL of NATAc extract

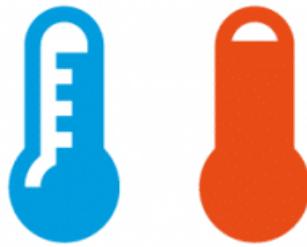
IPM: Isopropyl Myristate  
Cremophor EL: PEG-35 Castor Oil



# Stability studies

## Storage stability + 25°C and + 4°C:

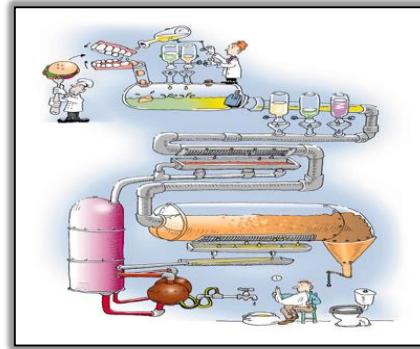
Physical stability: DLS and ELS, Chemical stability: HPLC-DAD analysis  
All formulations are physically and chemically stable for 8 weeks



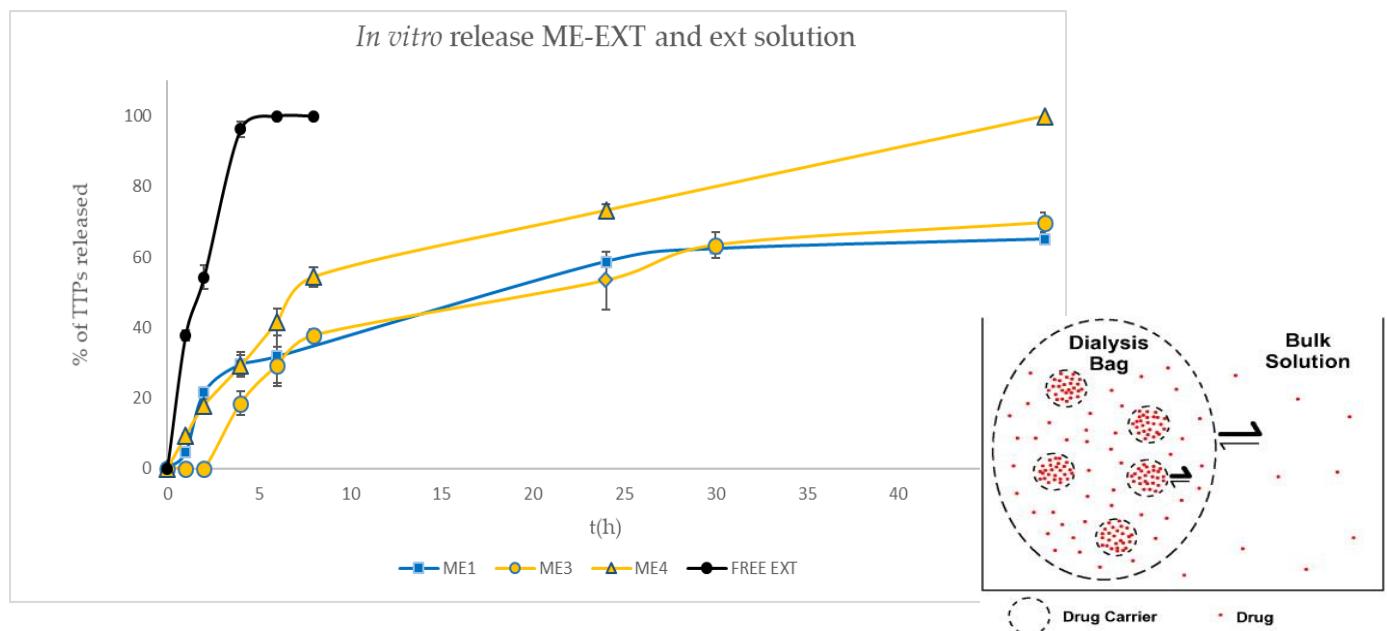
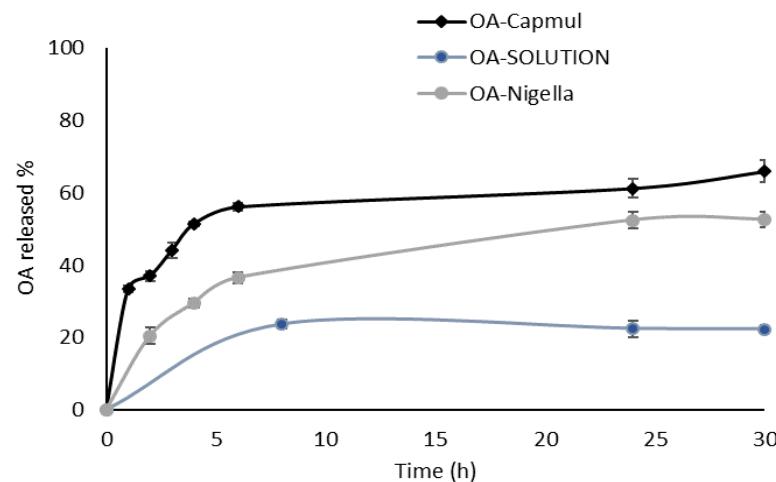
## Gastro-intestinal stability study in SGIF:

Physical stability: DLS and ELS.

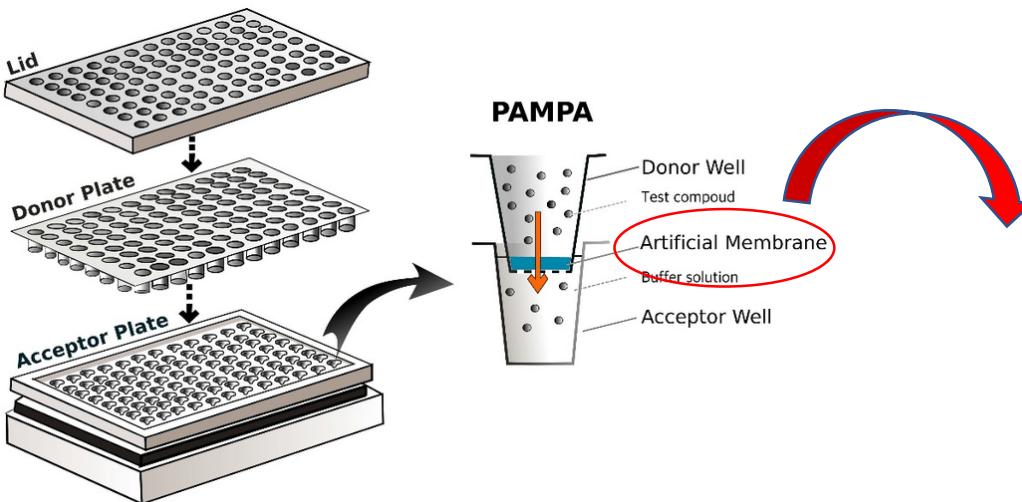
SGF, pH 1.2: 2 g of NaCl and 7 mL of HCl per liter of deionized water  
SIF, pH 6.8: 6.805 g of KH<sub>2</sub>PO<sub>4</sub> and 0.896 g of NaOH per liter of deionized water



# In vitro studies



# PAMPA parallel artificial membrane permeability assay



**Donor compartment:** formulation (diluted in PBS),  
or OA in SDS 0.5% w/v

**Acceptor compartment:** PBS/EtOH 95:5 or 70:30

Lecithin 10 mg/mL + Cholesterol 8 mg/mL in 1,7-octadiene (10 mL)

OA Pe  $2.7 \times 10^{-7} \pm 1.4 \times 10^{-8}$  cm/s

TTP70 extract (in PBS:EtOH 90:10) no permeation for 6 h. Recovery > 95%

ME 1-OA Pe  $5.7 \times 10^{-6} \pm 4.0 \times 10^{-7}$  cm/s at 1 h. Recovery 98%

ME1 TTP70 extract Pe TTPs  $4.31 \times 10^{-6} \pm 2.01 \times 10^{-7}$  cm/s at 1 h. Recovery 90%

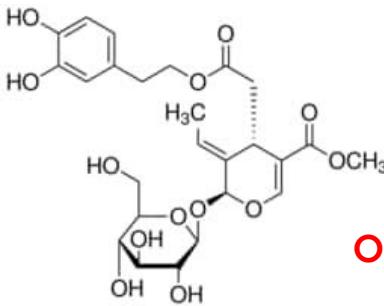
ME 2-OA Pe  $4.7 \pm 0.04 \times 10^{-5}$  cm/s at 1 h. Recovery 94%

ME3 TTP70 extract (1 mg/mL) Pe TTPs  $2.7 \times 10^{-6} \pm 3.70 \times 10^{-8}$  cm/s at 2h. Recovery 82%

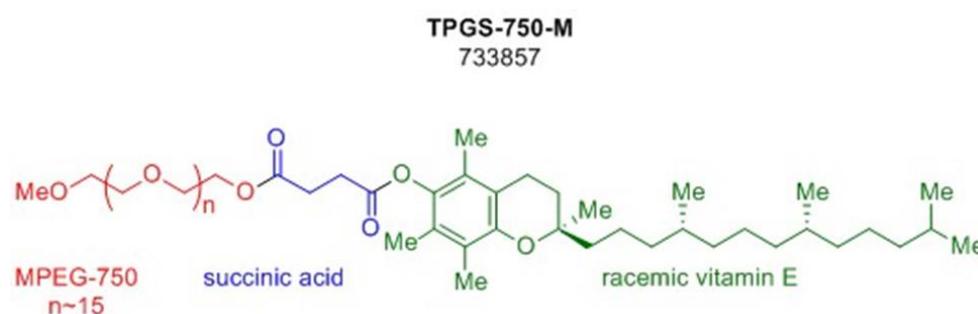
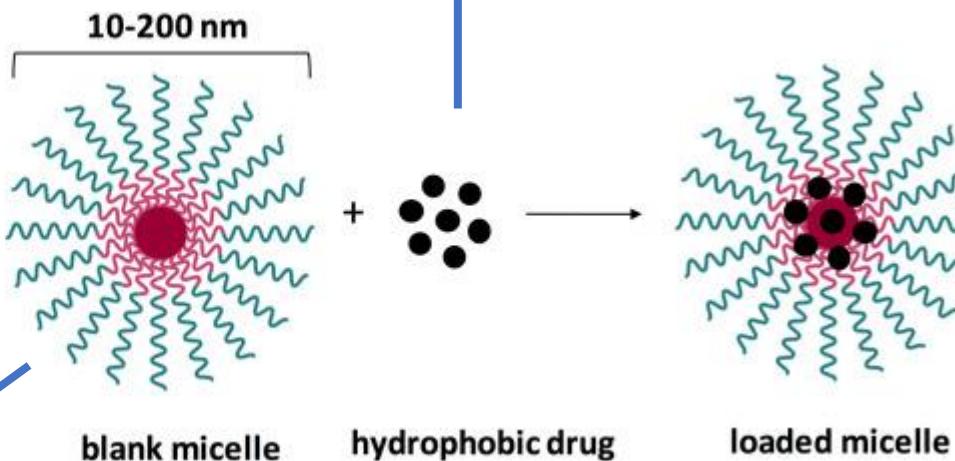
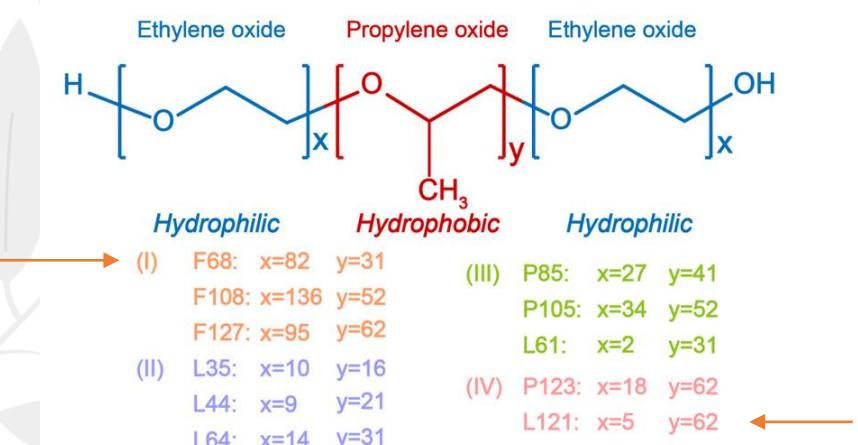
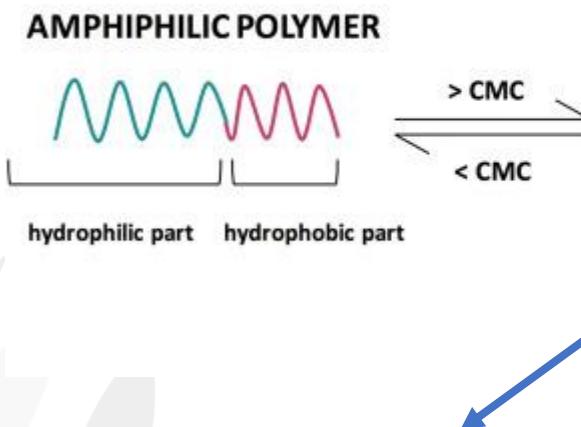
ME4 TTP70 extract (2 mg/mL) Pe TTPs  $1.2 \times 10^{-7} \pm 3.8 \times 10^{-8}$  cm/s at 2 h. Recovery 100%



# Polymeric micelles



**OLE40, polyphenols extract**



D- $\alpha$ -tocopheryl polyethylene succinate (Vitamin E TPGS or simply TPGS)



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# Pluronic L121: Pluronic P188: TPGS 2:2:1

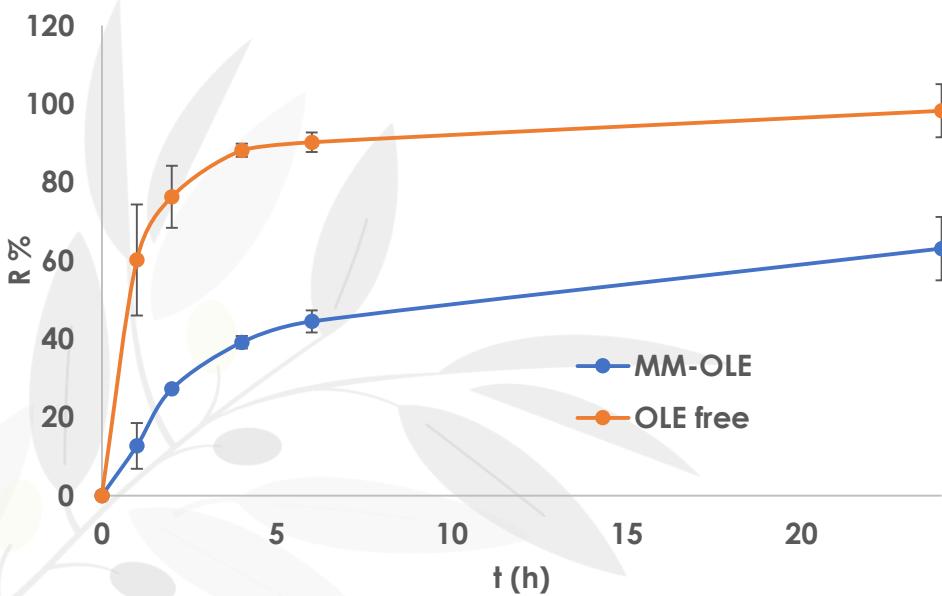
- L121 200 mg
- P188 200 mg
- TPGS 100 mg
- OLE 15 mg/mL

**Micelles**  
Size 14.21+/- 0.14  
PdI 0.19 +/-0.05  
**Recovery % :** 91.55 +/- 4.55  
**EE% :** 65.82% +/-1.11

**Micelles after freeze-drying process**  
Size 15.97 +/- 1.13  
PdI 0.24 +/-0.04  
**Recovery % :** 88.12 +/- 2.07  
**EE% :** 64.71% +/- 2.82

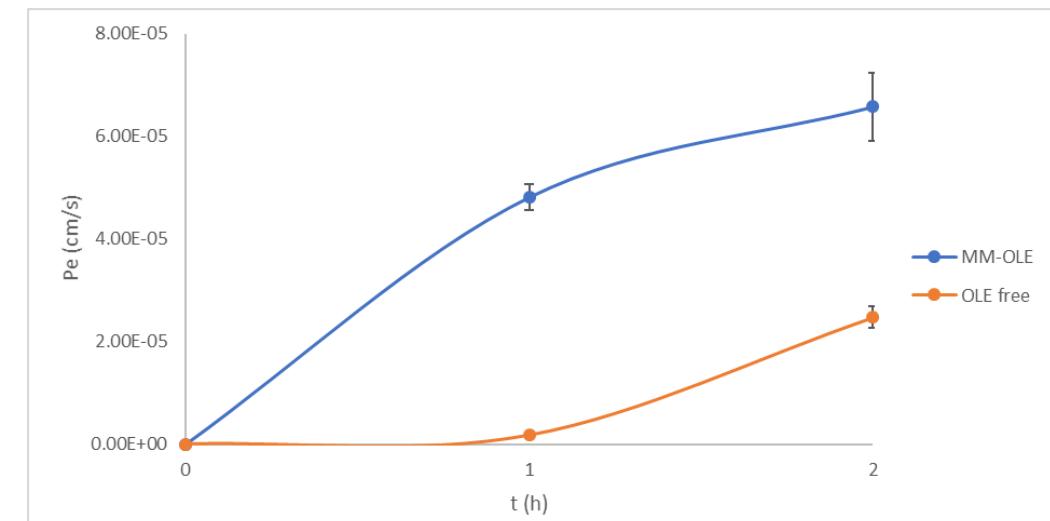
## In vitro release study

Release medium: PBS and tween 80 (0.5% w/v)



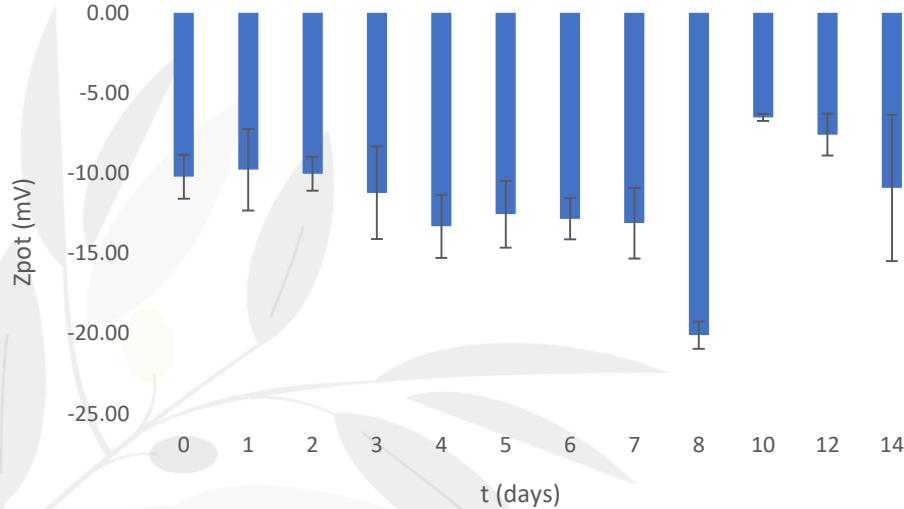
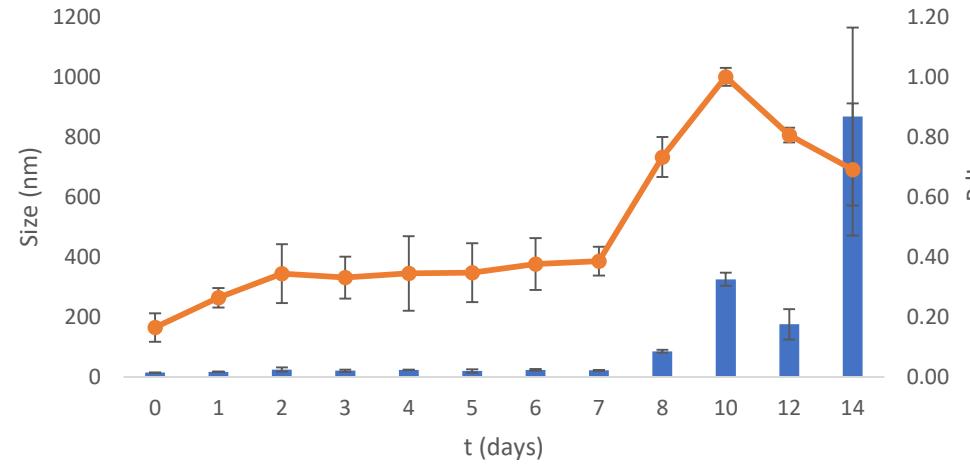
## PAMPA

Donor compartment: formulation or OLE inPBS, Acceptor compartment: PBS



# Stability studies

## Colloidal dispersion

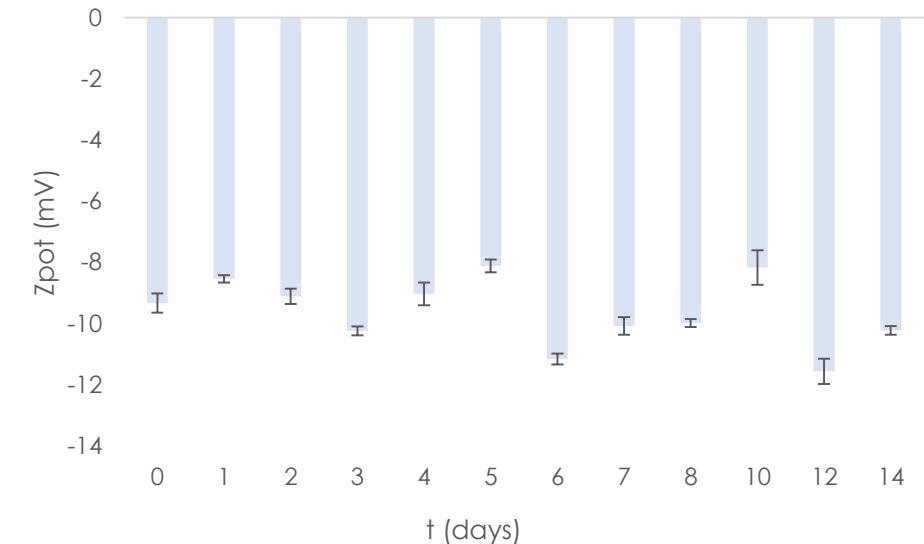
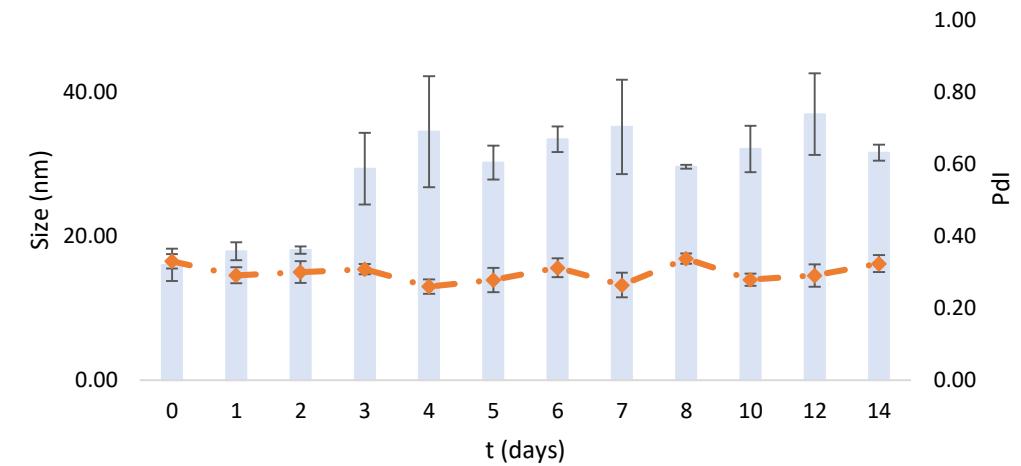


days	EE%	ds
1	68.24	2.56
7	60.39	4.78
14	57.34	3.65

## Storage stability at + 4°C:

Physical stability: DLS and ELS, Chemical stability: HPLC-DAD analysis

## Freeze-dried product

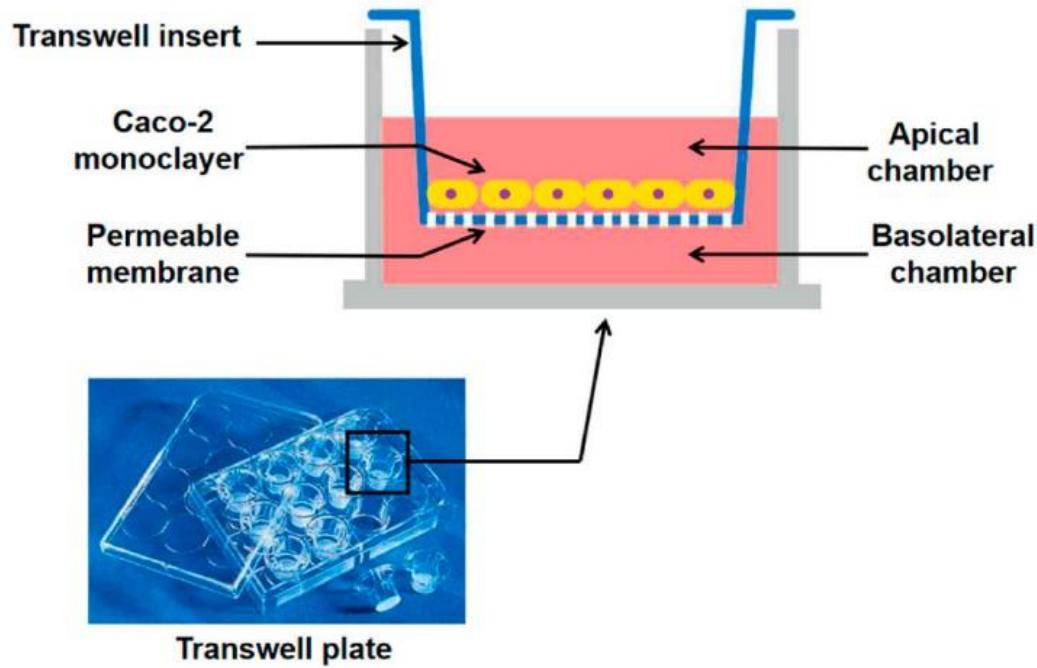


days	EE%	ds
1	63.45	0.98
7	61.38	2.42
14	61.12	3.18



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# Caco-2 cells permeability test



Papp OLE 40       $1.10 \pm 0.03 \times 10^{-6} \text{ cm/s}$   
Papp PM OLE 40     $2.18 \pm 0.18 \times 10^{-6} \text{ cm/s}$

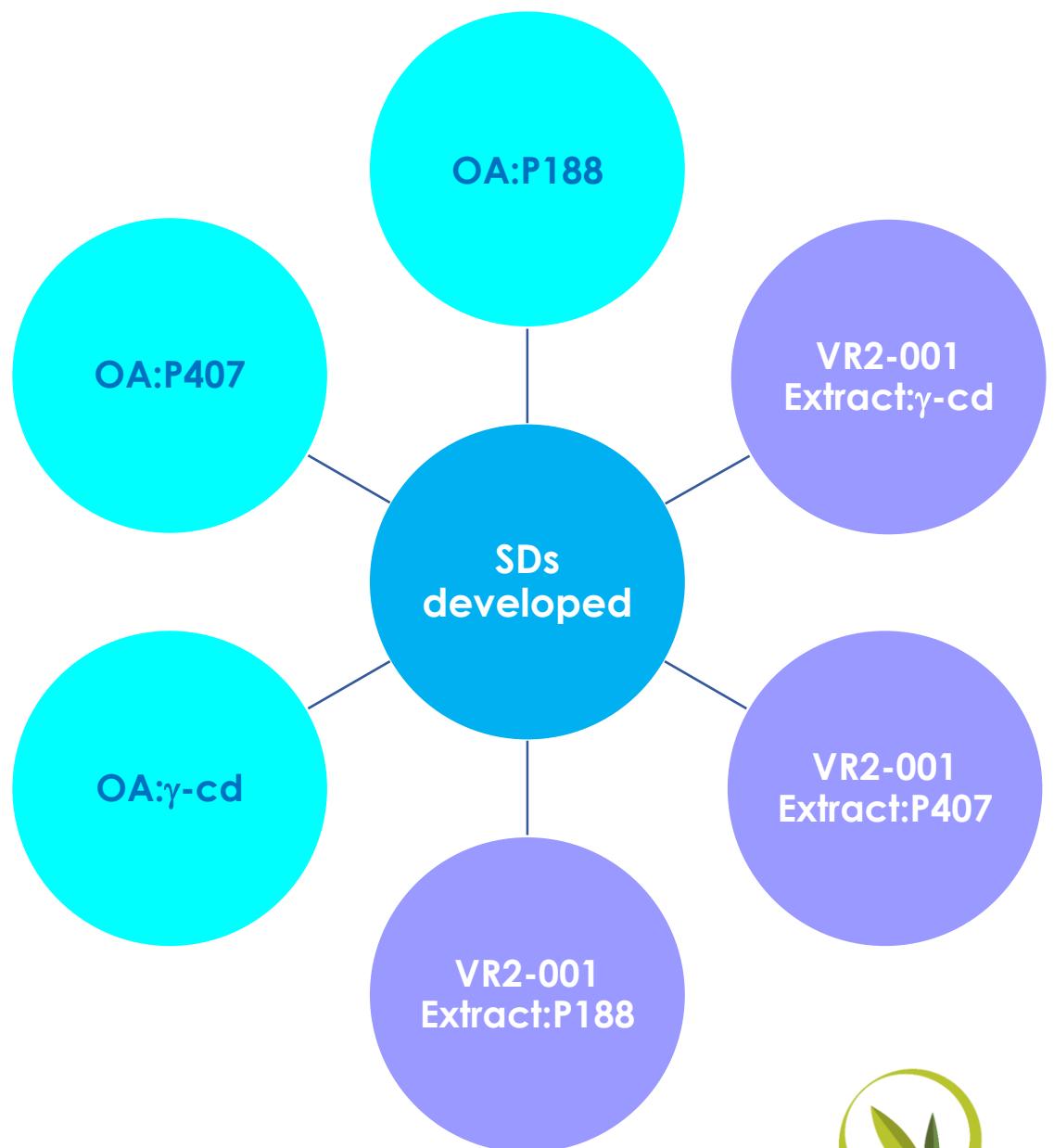
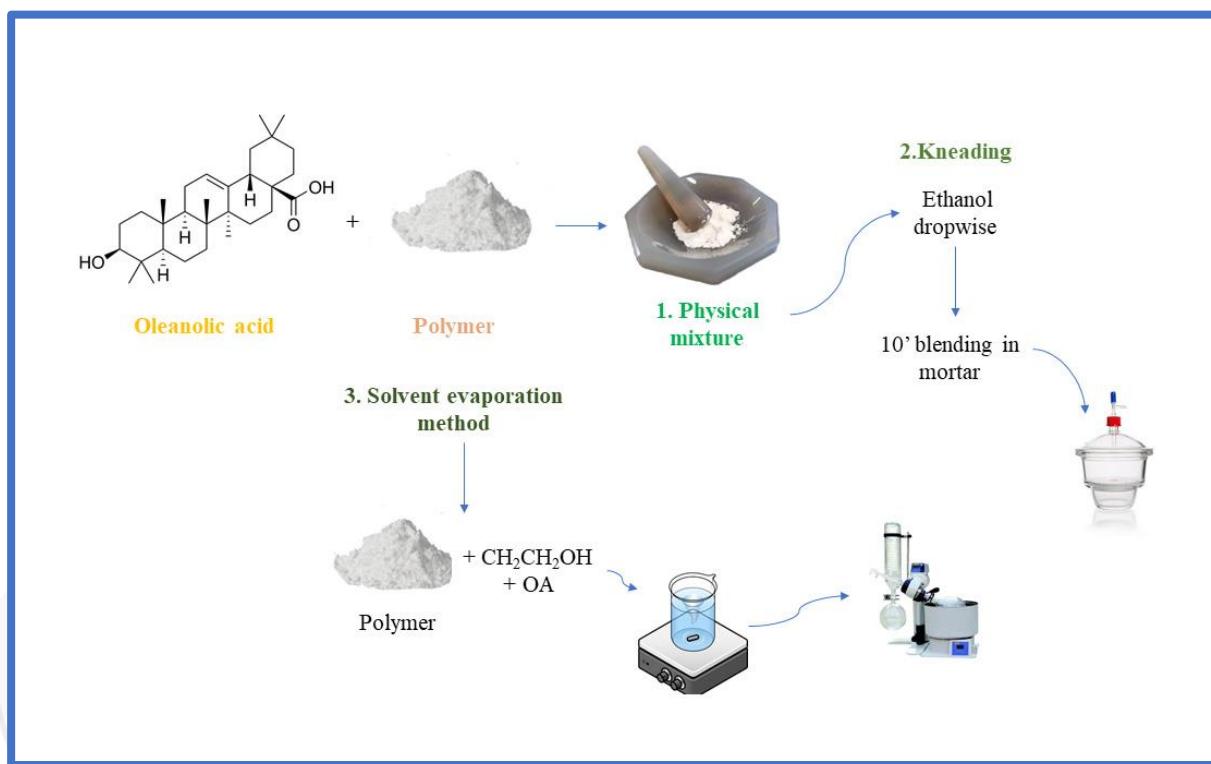
Recovery 86-90%  
P% (LY) 1.30 and 1.15%



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# Solid dispersions

Solid dispersion is the dispersion of one hydrophobic active ingredient in an inert hydrophilic carrier in the solid state prepared using different methods. In these systems, the drug is combined with a water-soluble polymer to produce a single-phase amorphous mixture of the drug and the polymer.



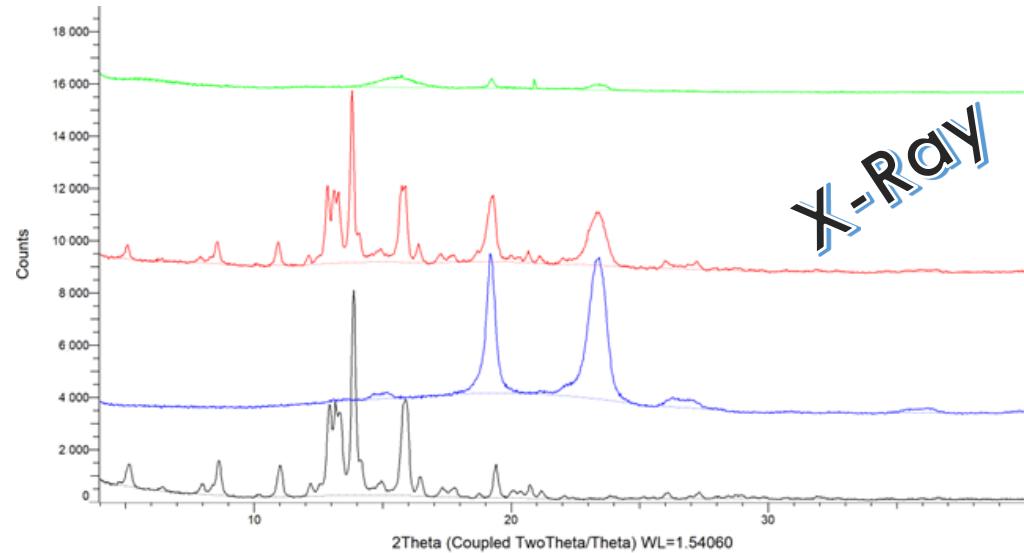
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# SD's characterization

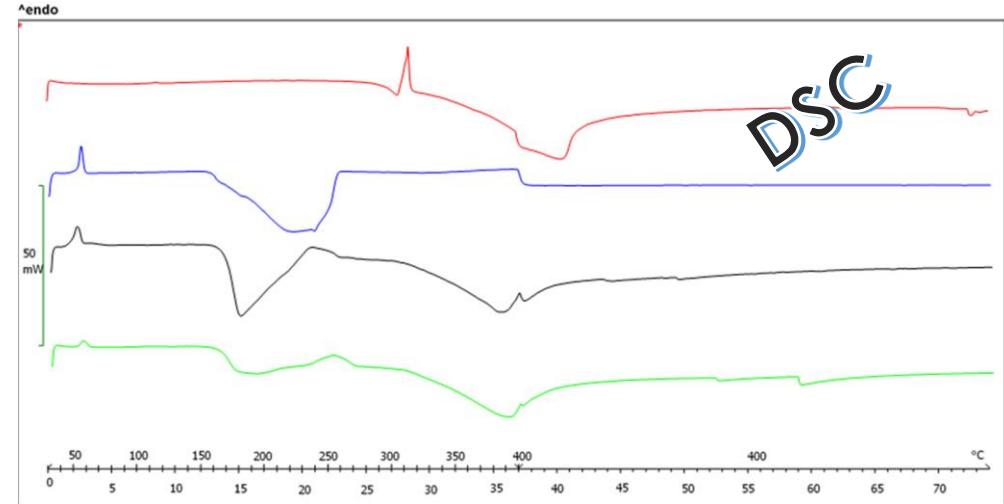


**Table 2.** Solubility of OA in hydrophilic polymers, using 1:2 drug-to-polymer weight ratio and different preparation methods (PM: physical mixture; K: kneading; SEM: solvent evaporation method).

Method	Polymer	Solubility ( $\mu\text{g/mL}$ )
PM	Poloxamer 188	14 $\pm$ 2
	Poloxamer 407	32 $\pm$ 0.7
	PEG 4000	6 $\pm$ 0.8
	PEG 6000	5 $\pm$ 0.8
K	$\gamma$ -CD	66 $\pm$ 3
	Poloxamer 188	79 $\pm$ 1
	Poloxamer 407	130 $\pm$ 7
	PEG 4000	12 $\pm$ 2
SEM	PEG 6000	9 $\pm$ 2
	$\gamma$ -CD	7 $\pm$ 0
	Poloxamer 188	190 $\pm$ 42
	Poloxamer 407	170 $\pm$ 28
	PEG 4000	10 $\pm$ 1
	PEG 6000	12 $\pm$ 3
	$\gamma$ -CD	145 $\pm$ 4



OA (black), Poloxamer 407 (blue), PM1:1 (red), and SEM1:1 (green).

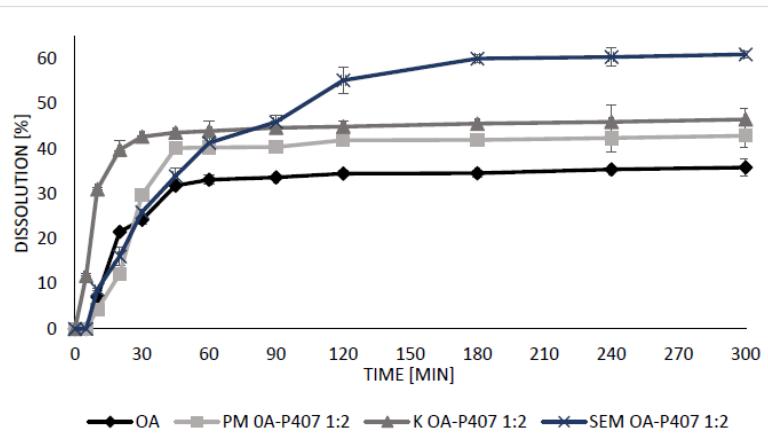


OA (red), Poloxamer 407 (blue), PM OA-P407 1:2 (black), and SEM OA-P407 1:2 (green)

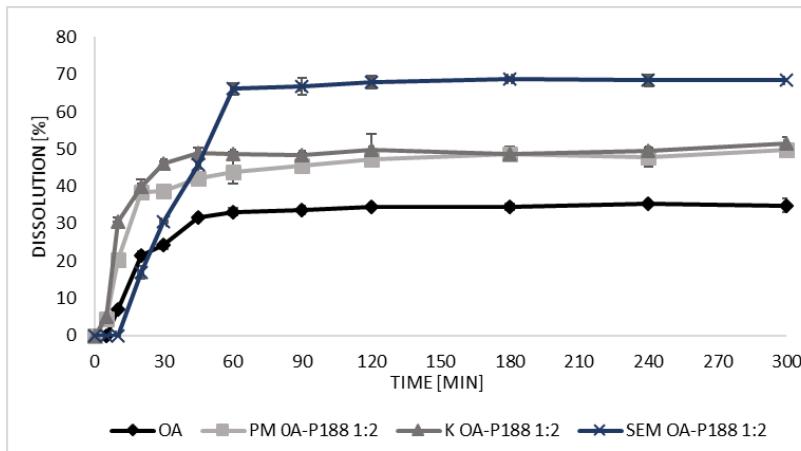
# OA SD Dissolution test



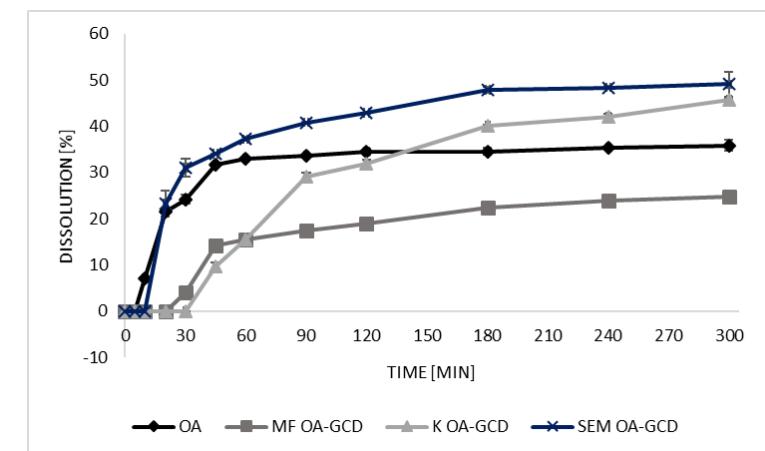
Poloxamer 407



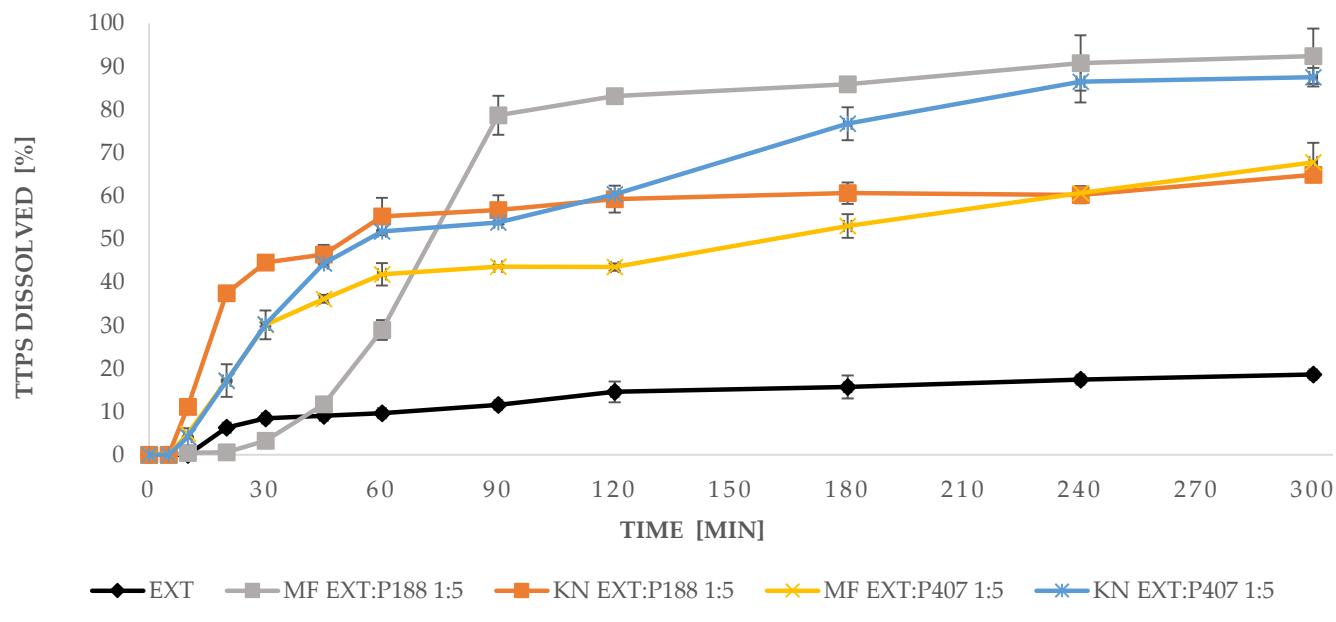
Poloxamer 188



$\gamma$ -CD

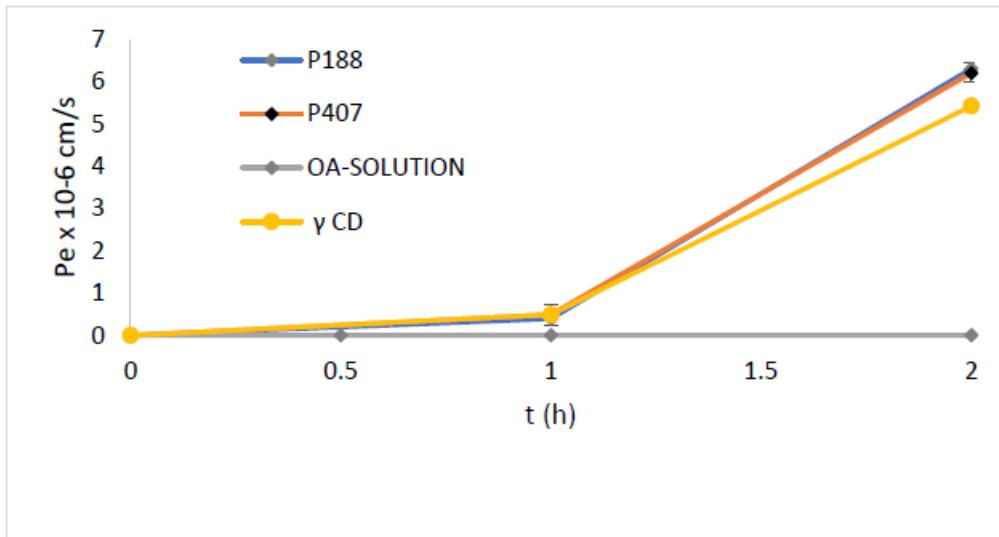
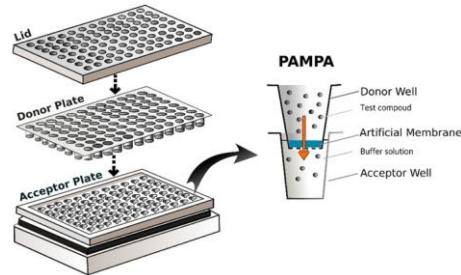


# TTP70 SD Dissolution test



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# PAMPA



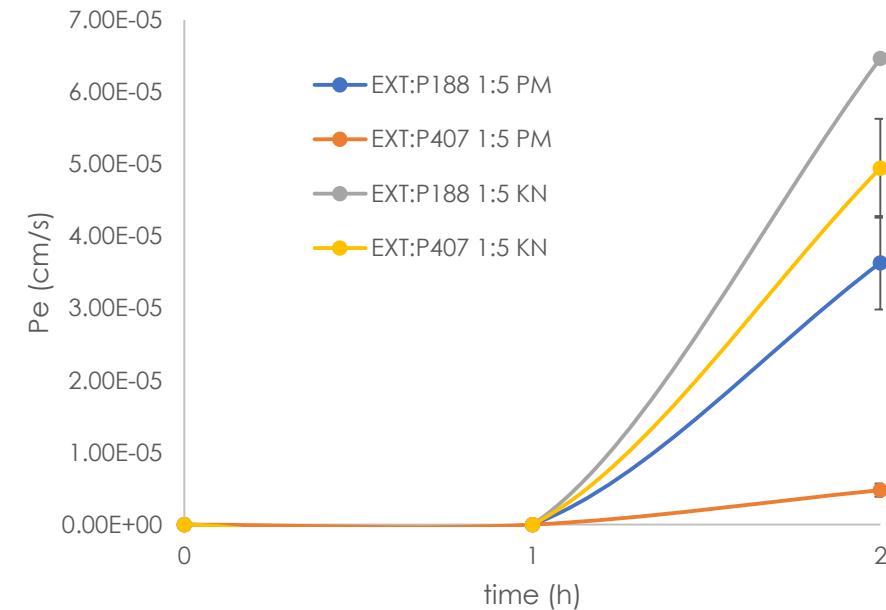
OA Pe  $2.7 \times 10^{-7} \pm 1.4 \times 10^{-8}$  cm/s

P407 Pe  $6.2 \pm 0.22 \times 10^{-5}$  cm/s

P188 Pe  $6.3 \pm 0.53 \times 10^{-5}$  cm/s

$\gamma$ -CD Pe  $5.43 \pm 0.12 \times 10^{-5}$  cm/s

Recovery 85%



TTPs extract no permeation for 6 h.

MF P407 Pe  $4.78 \times 10^{-6} \pm 9.16 \times 10^{-7}$  cm/s

KN P407 Pe  $4.90 \times 10^{-5} \pm 6.87 \times 10^{-6}$  cm/s

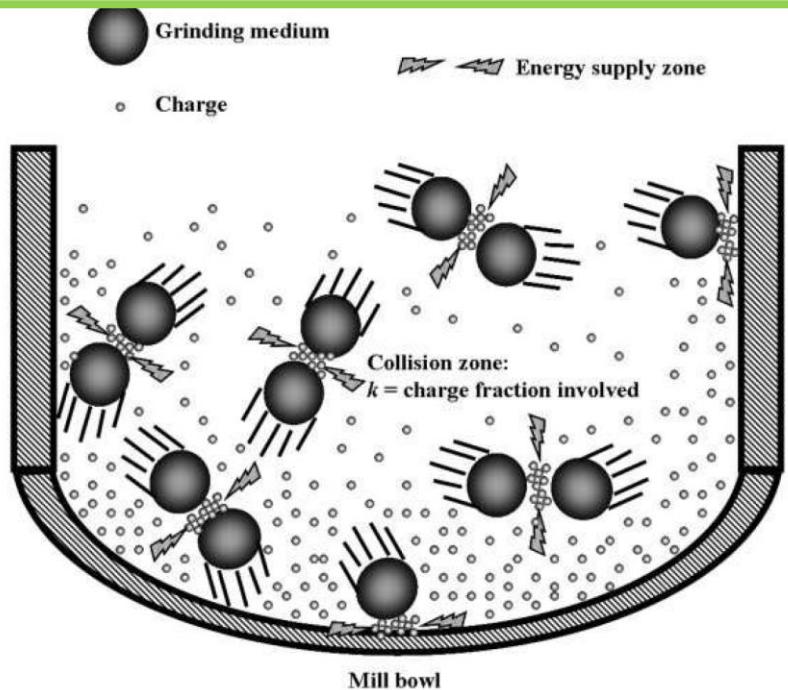
MF P188 Pe  $3.63 \times 10^{-5} \pm 6.48 \times 10^{-6}$  cm/s

KN P188 Pe  $6.46 \times 10^{-5} \pm 6.56 \times 10^{-6}$  cm/s

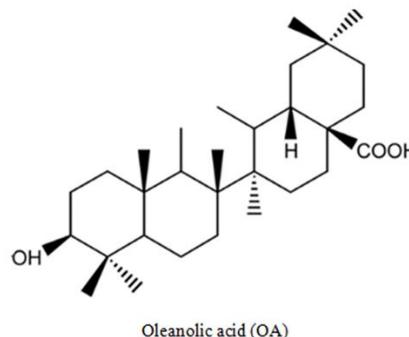
Recovery > 95%



# Mechanochemical activation



**Figure 1.** Schematic representation of what happens inside mill bowl during grinding. Grinding media transfer mechanical energy to the charge (solid drug) in pulse form as, in each collision, only a small fraction  $k$  of the total charge is involved.



**Stabilizer  
Na Cholate**



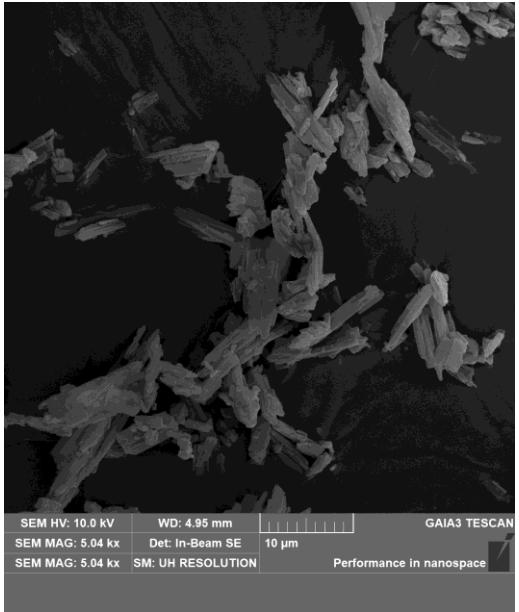
Model: Retsch MM200  
Tubes: 25 ml in stainless steel  
Sphere: Tempered steel 2x12 mm + 2x5 mm

The mill is the typical energy supplier and grinding and cogrinding are the common processes inducing mechanochemical transformations. There is a complete amorphization of the drug.

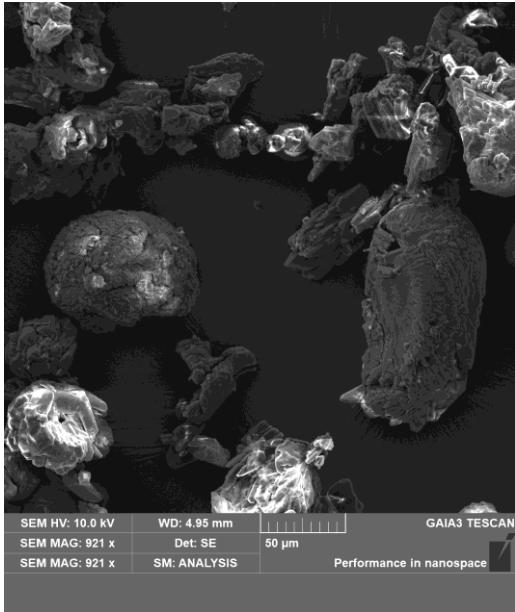


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# Results: SEM

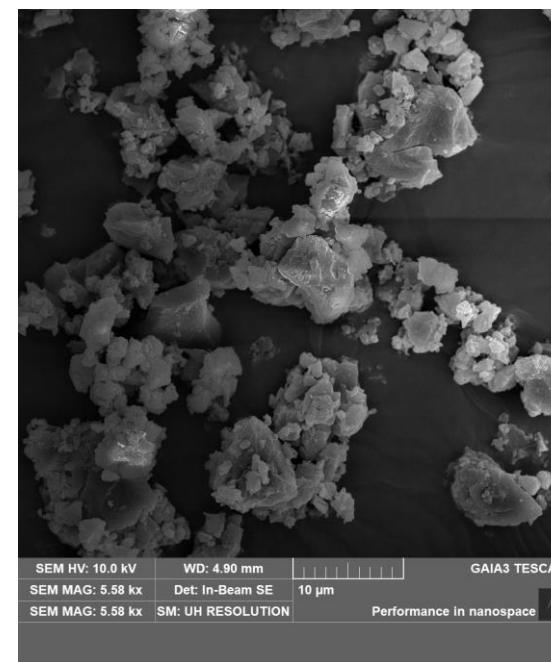
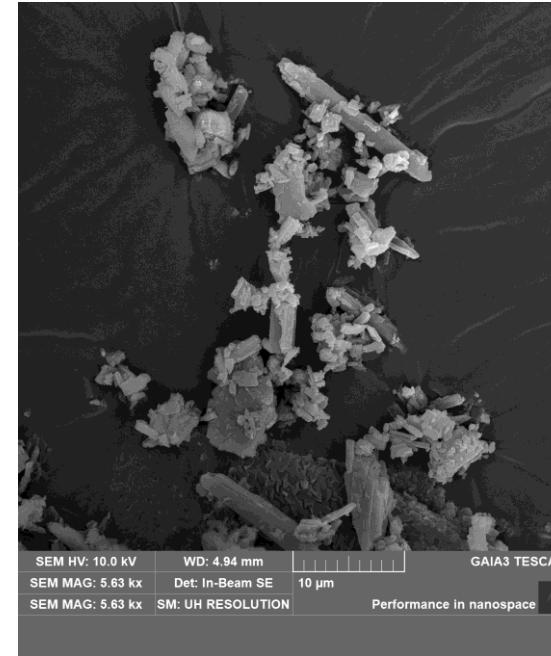


OA



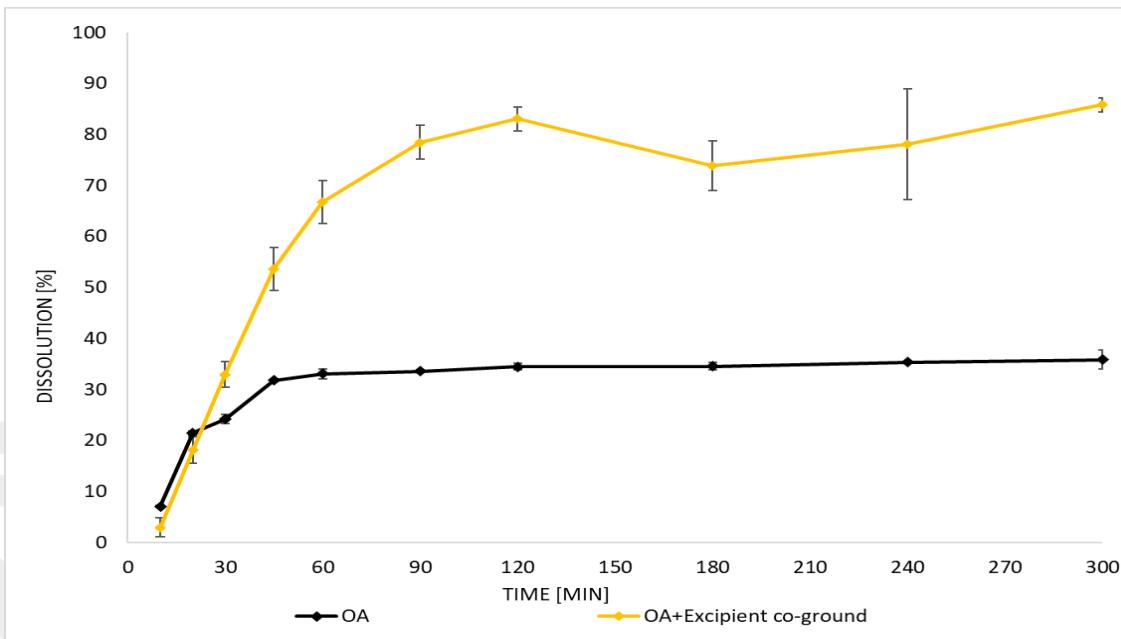
Na cholate

PM  
Coground



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# Dissolution test



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# Conclusions

- ✓ Micro/Nanoformulations were developed to improve technological performance of olive bioactive compounds.
- ✓ 6 MEs, 6 SDs, 2Co-G for triterpenes and 1 PM for polyphenolic extract.
- ✓ The optimized formulations improved the solubility and *in vitro* permeability of olive bioactive compounds and olive leaf fractions.
- ✓ The formulations improved the bioactivity of compounds in *in vitro* tests.





Thanks for your kind attention!



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